

Laboratory Investigations for CO₂ Based EOR

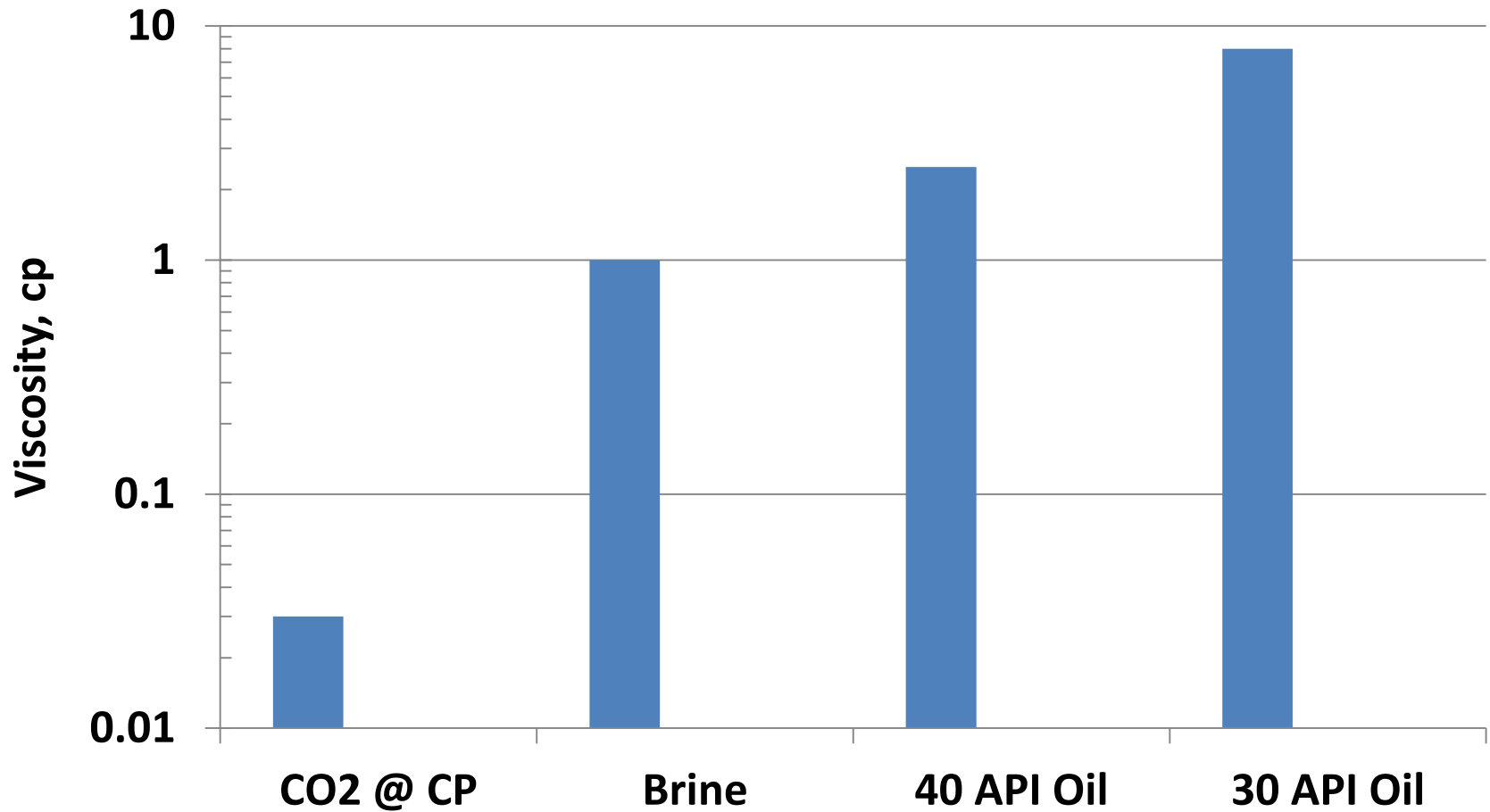
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Rice University
CO₂ for EOR as CCUS
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Challenges Facing CO₂ EOR

- **Availability of CO₂**
- **Minimum miscibility pressure (MMP)**
- **Asphaltene deposition**
- **Mobility control**
 - **Low viscosity of CO₂, viscous fingering**
 - **WAG**
 - **Gravity segregation**
 - **Heterogeneities**
 - **Foam mobility control**
- **Adsorption**

Viscosity of CO₂ and Reservoir Fluids



Displacement Fronts for Different Mobility Ratios and Injected Pore Volumes

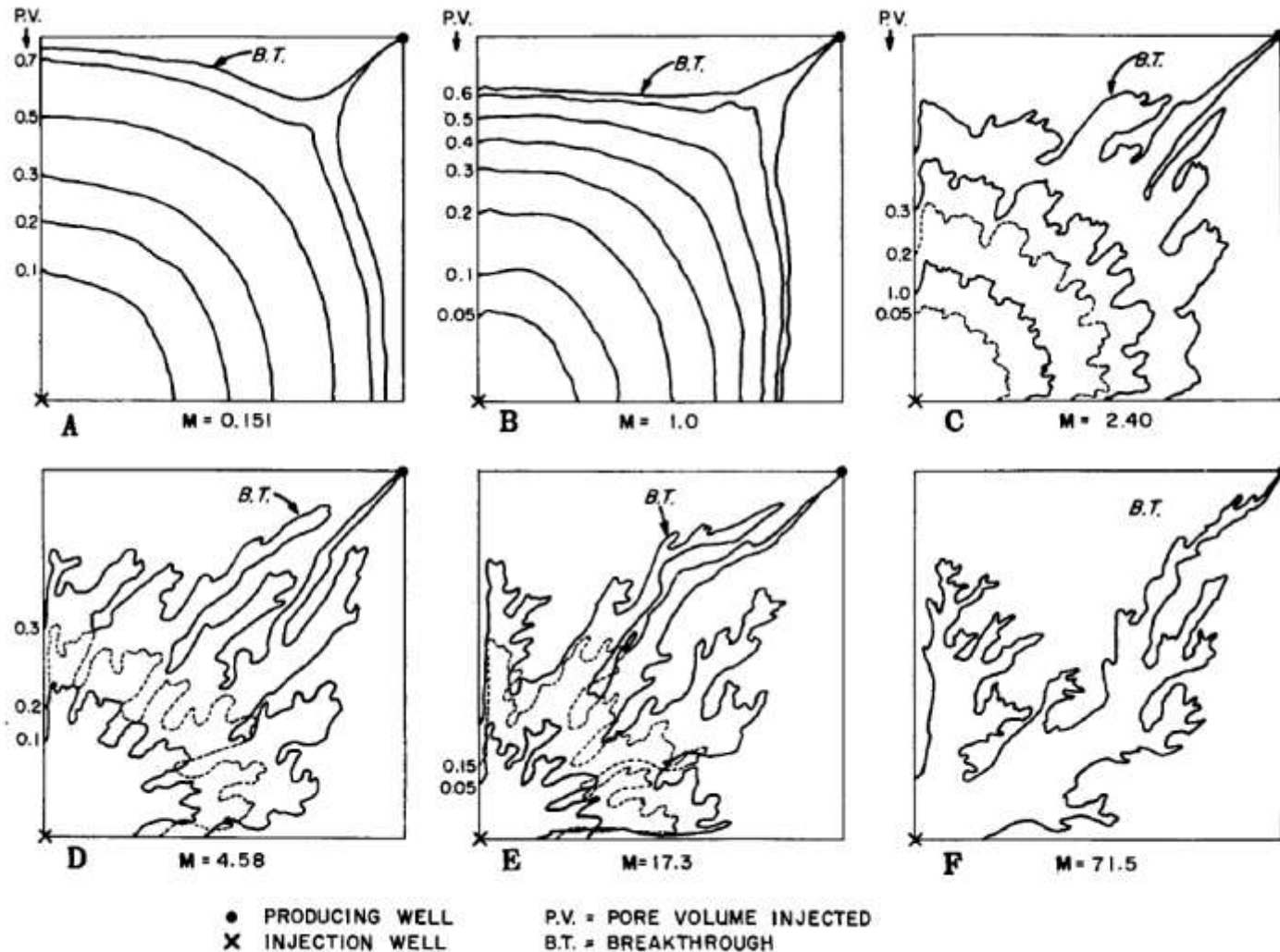
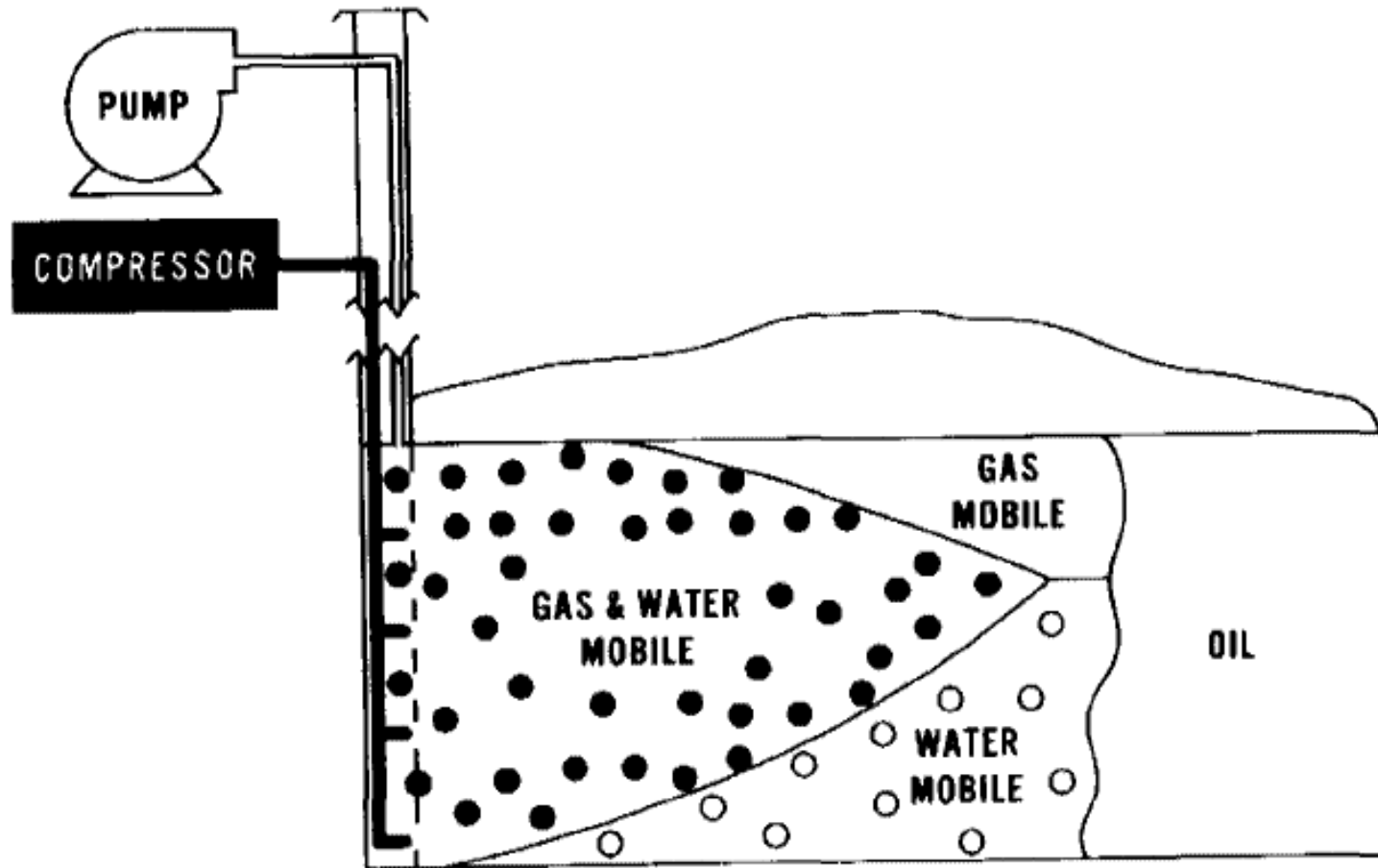
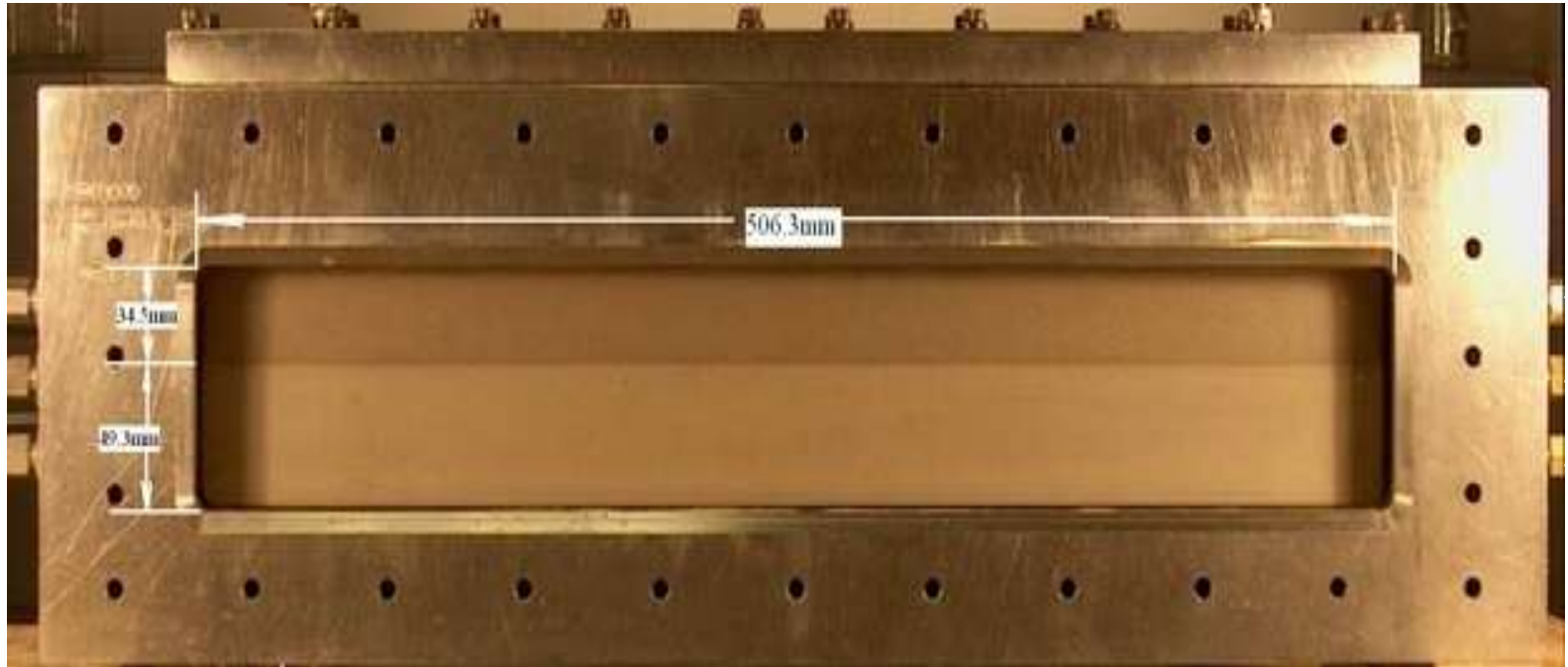


FIG. 5—DISPLACEMENT FRONTS FOR DIFFERENT MOBILITY RATIOS AND INJECTED PORE VOLUMES UNTIL BREAKTHROUGH, QUARTER OF A FIVE-SPOT.

Gravity Segregation with WAG



What is the effect of geological heterogeneity on EOR?



Layered sandpack model with 20:1 permeability Contrast

Waterflood and WAG

Waterflood or WAG, 4 psi; Sweep is only a function of liquid injected

$f_g = 0$ water only



0.0 TPV



1.0 TPV



2.0 TPV



3.0 TPV



4.0 TPV



5.0 TPV



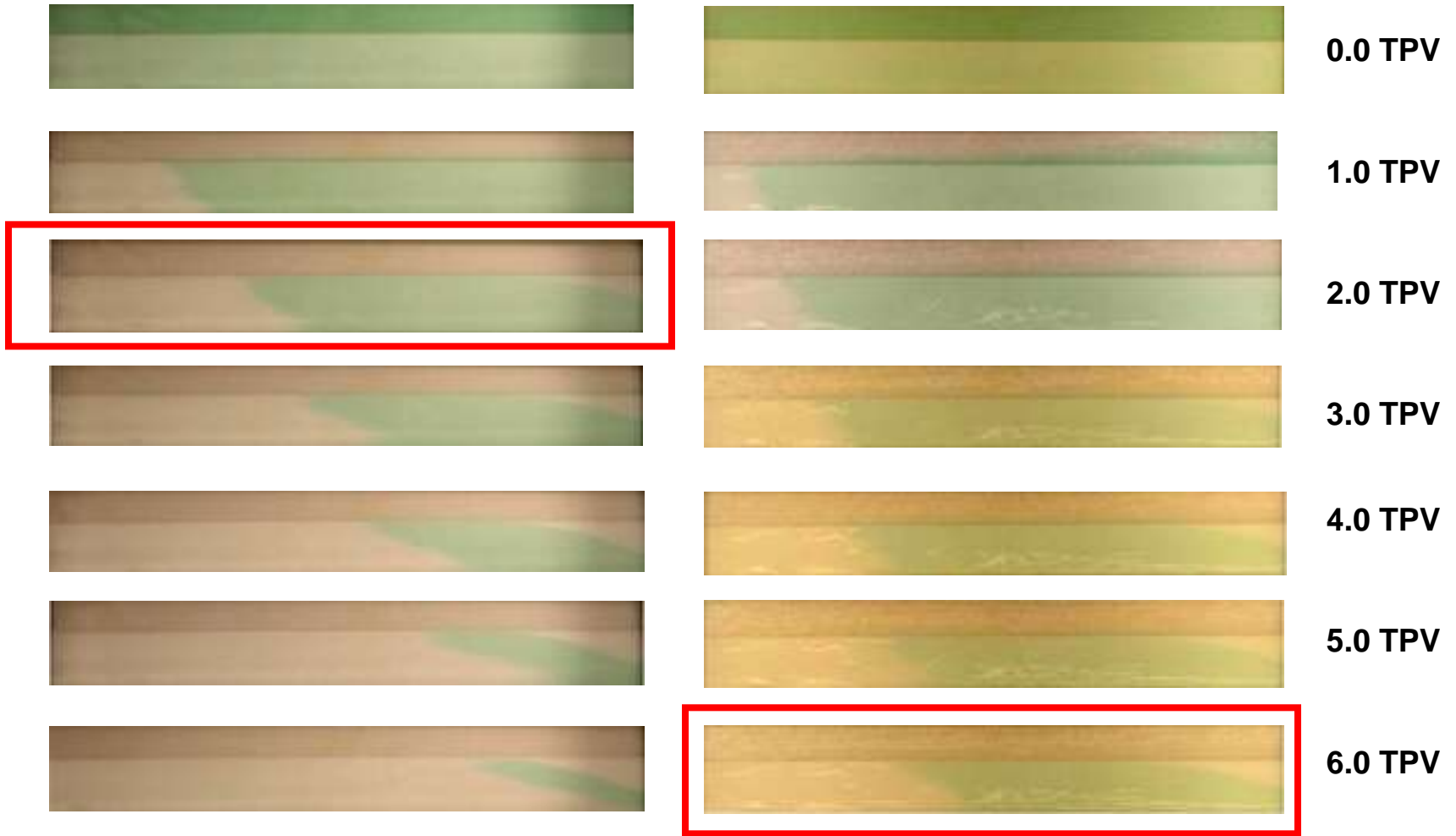
6.0 TPV

Waterflood and WAG

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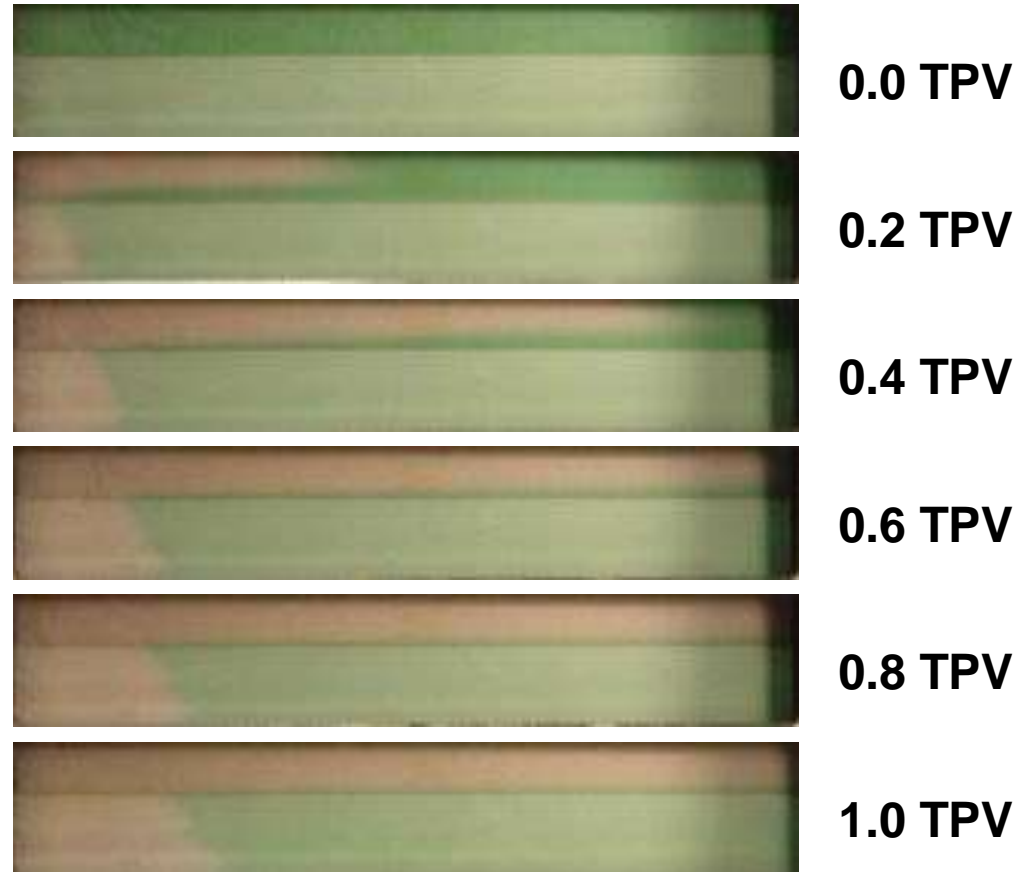
$fg = 0$ water only

WAG $fg = 2/3$



Layered sandpack with 19:1 permeability contrast about half-swept with water only but about completely swept with surfactant-alternated-gas (SAG)

Water only, 4 psi



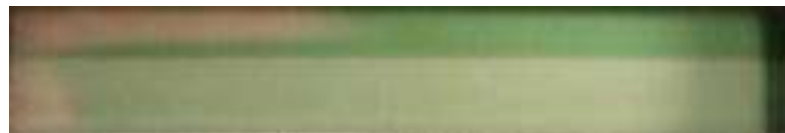
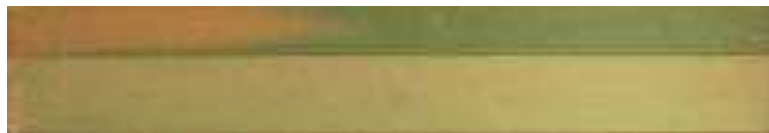
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SAG, 6 psi, $f_g=1/3$

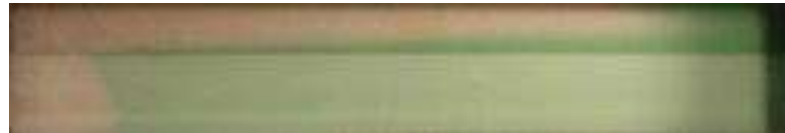
Water only, 4 psi



0.0 TPV



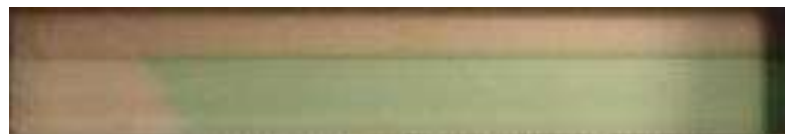
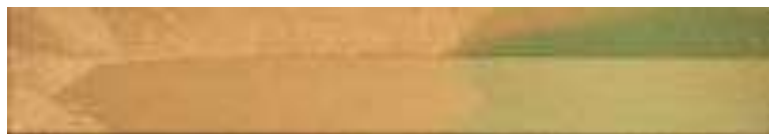
0.2 TPV



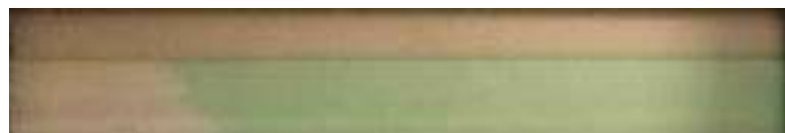
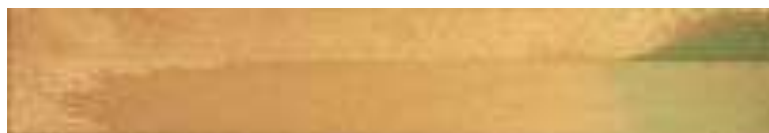
0.4 TPV



0.6 TPV

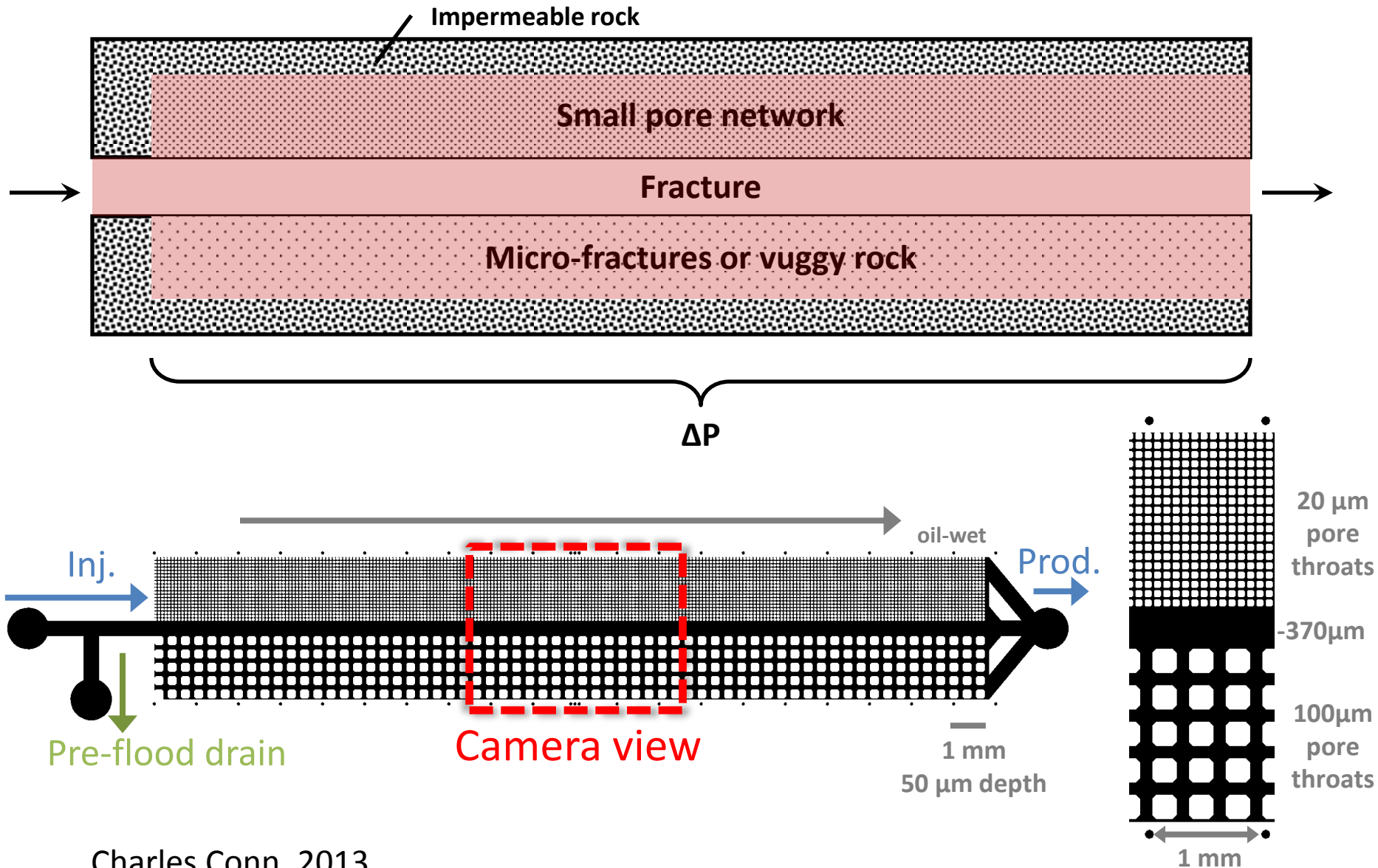


0.8 TPV



1.0 TPV

How does foam displace oil from heterogeneous, oil-wet, fractured system?



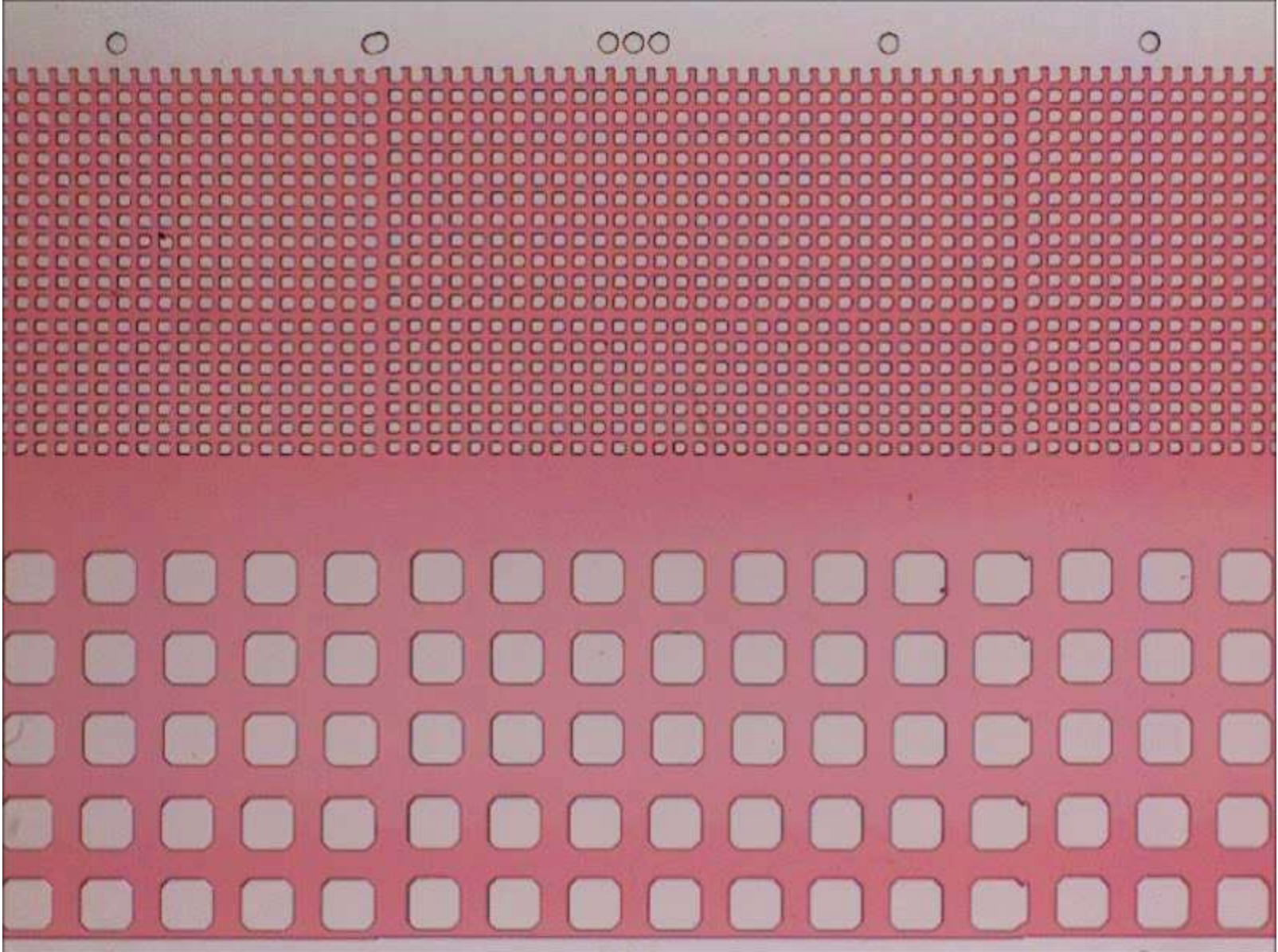
CONTROL

Air & Water

no surfactant

<https://drive.google.com/file/d/0BwZmYz9LHrmoSHFZc1Y2ZkF3QzA/edit?usp=sharing>

Water 0.05 mL/hr, Air 150 mbar

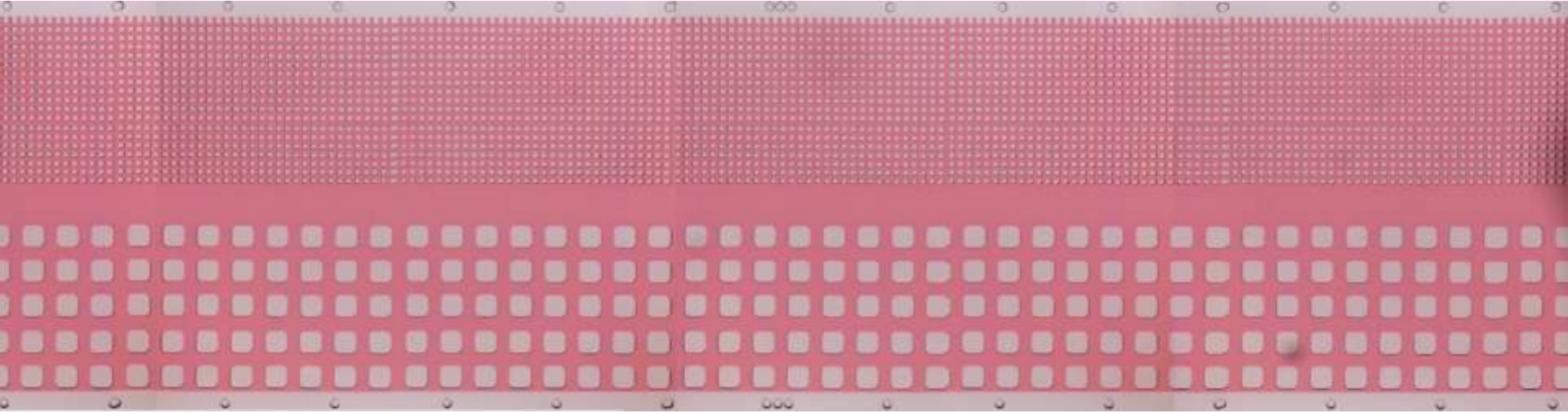


CONTROL

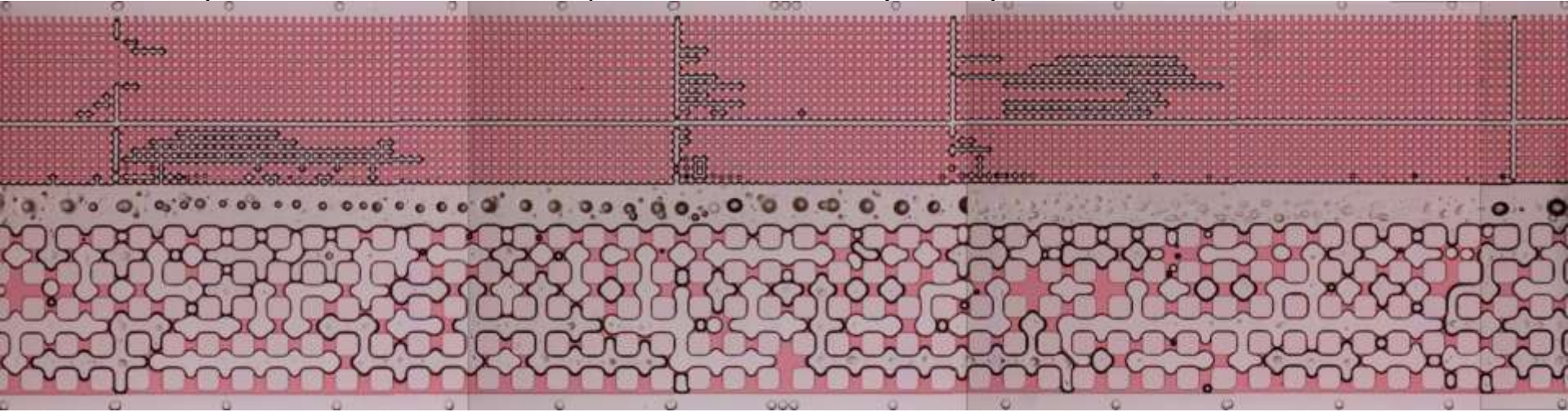
Air & Water

no surfactant

Pre-flood (pink = paraffin oil)



Post-flood (thick dark lamellae = air, thin lamellae = aqueous)



BASE CASE

1% AOS:LB, Seawater ionic strength (NaCl)

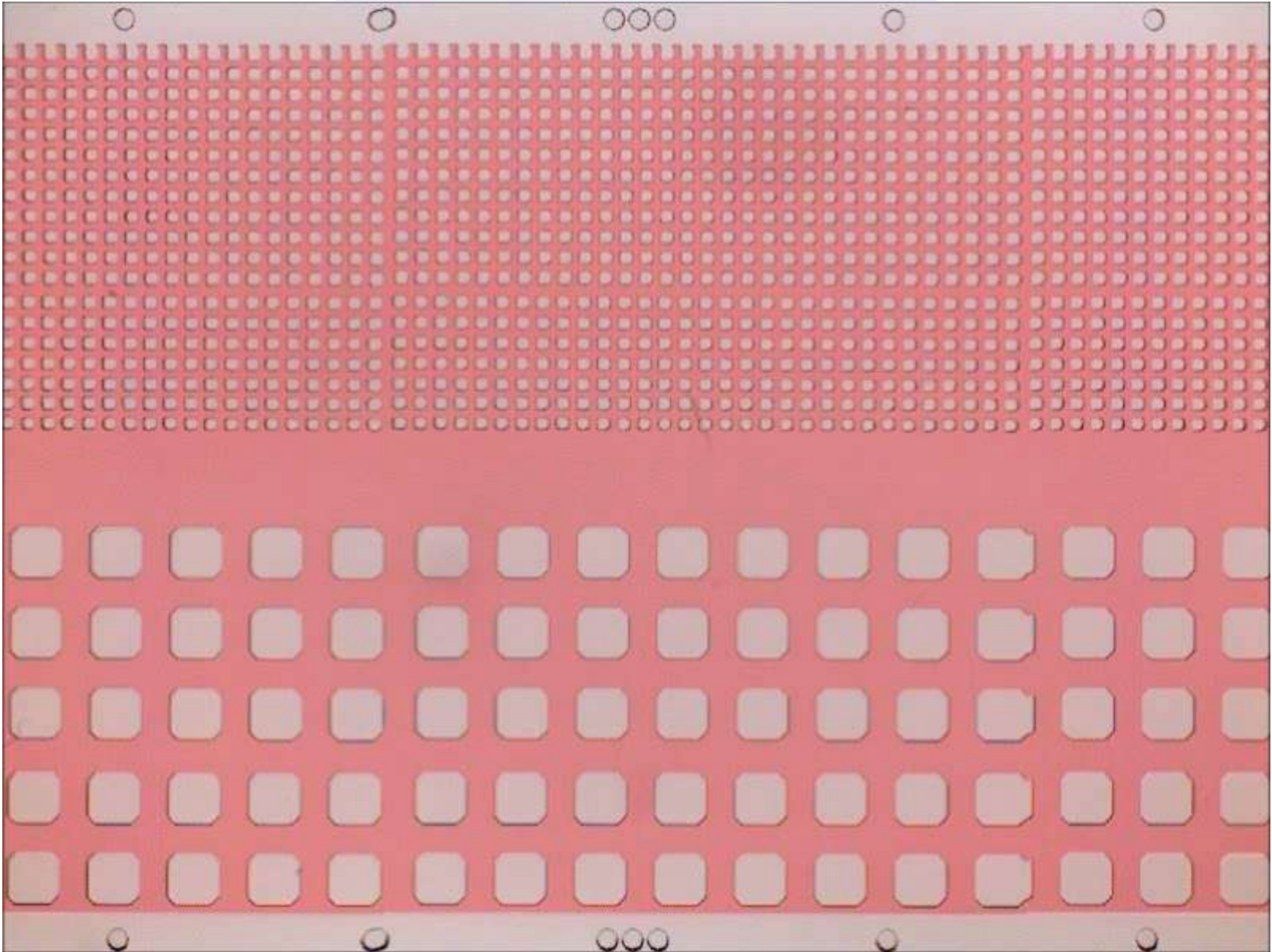
AOS₁₄₋₁₆ : LB

1 : 1

(good foamer)

Surfactant 0.05 mL/hr, Air 150 mbar

<https://drive.google.com/file/d/0BwZmYz9LHrmoa1dwcVJDZDNHcEk/edit?usp=sharing>

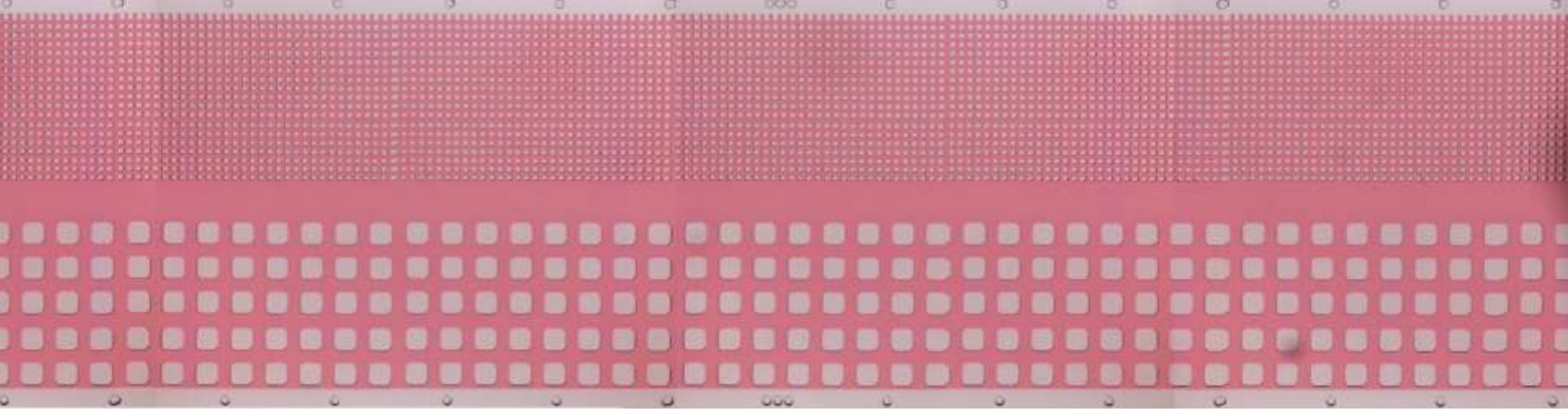


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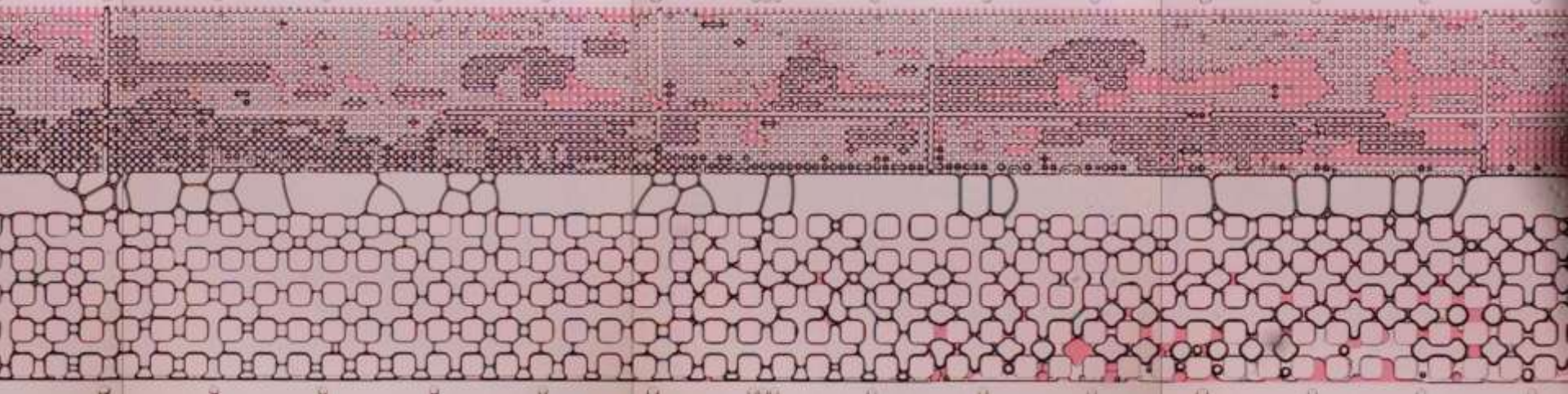
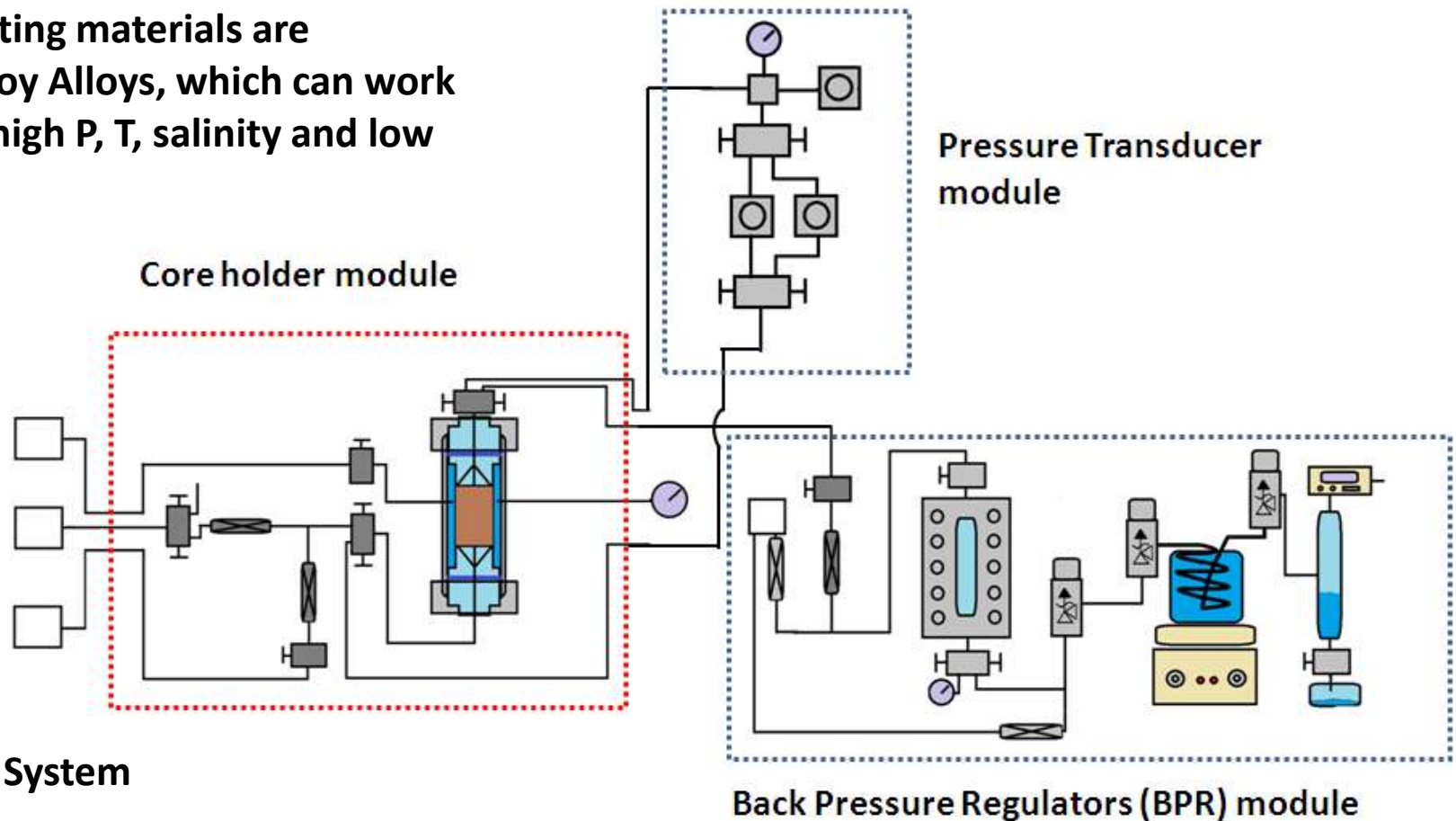
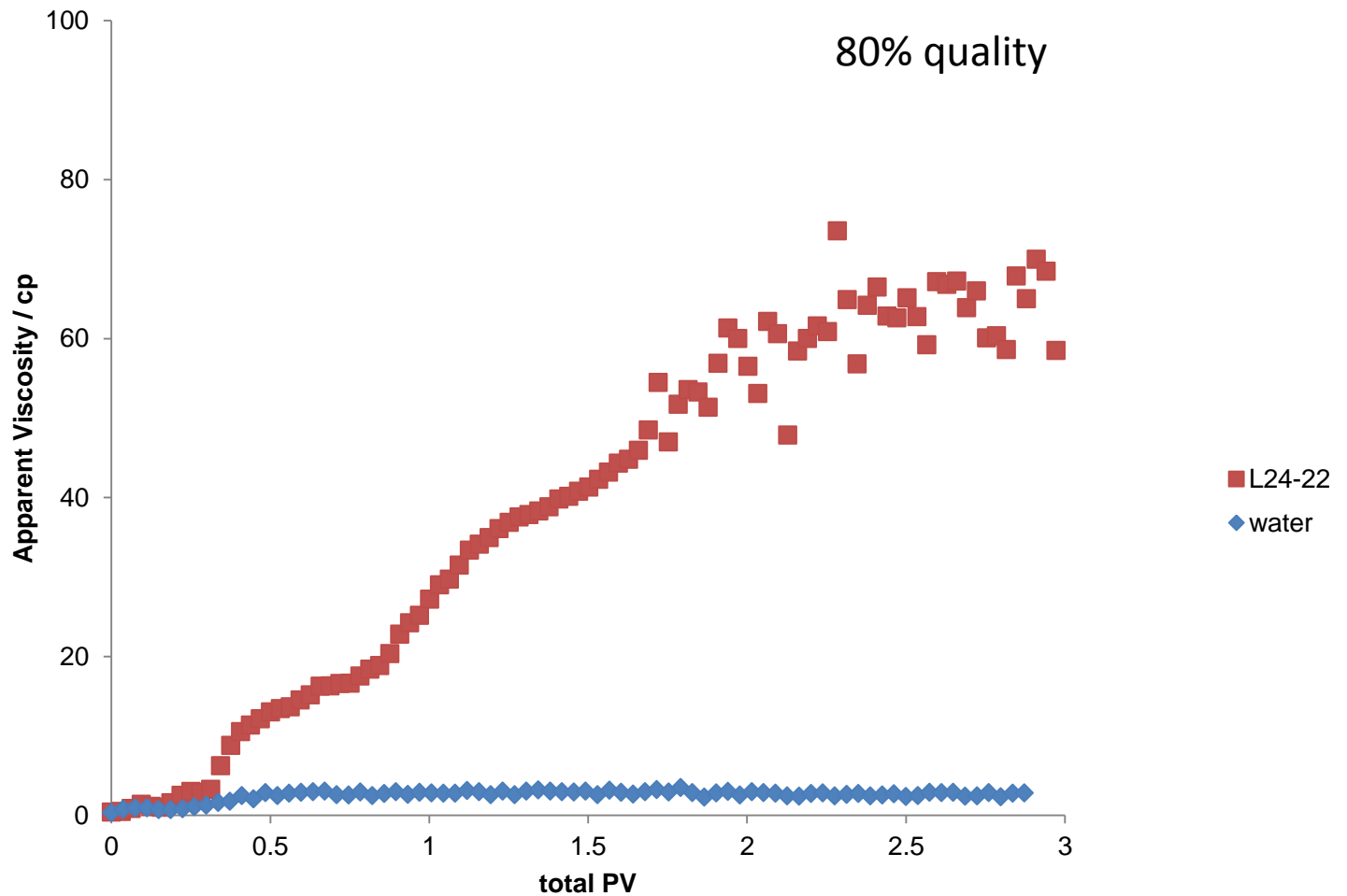


Diagram of the high temperature and high pressure core flooding setup

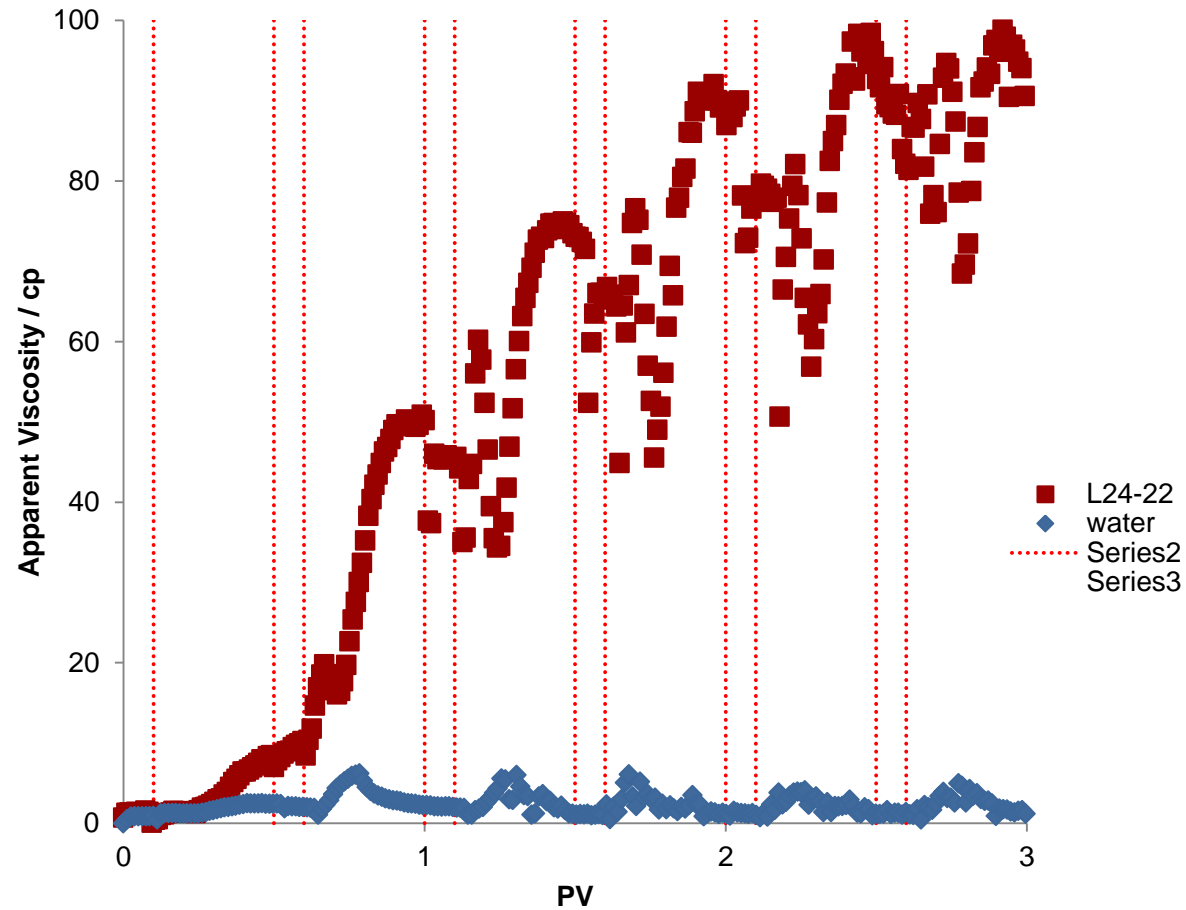
All wetting materials are Hastelloy Alloys, which can work under high P, T, salinity and low pH.



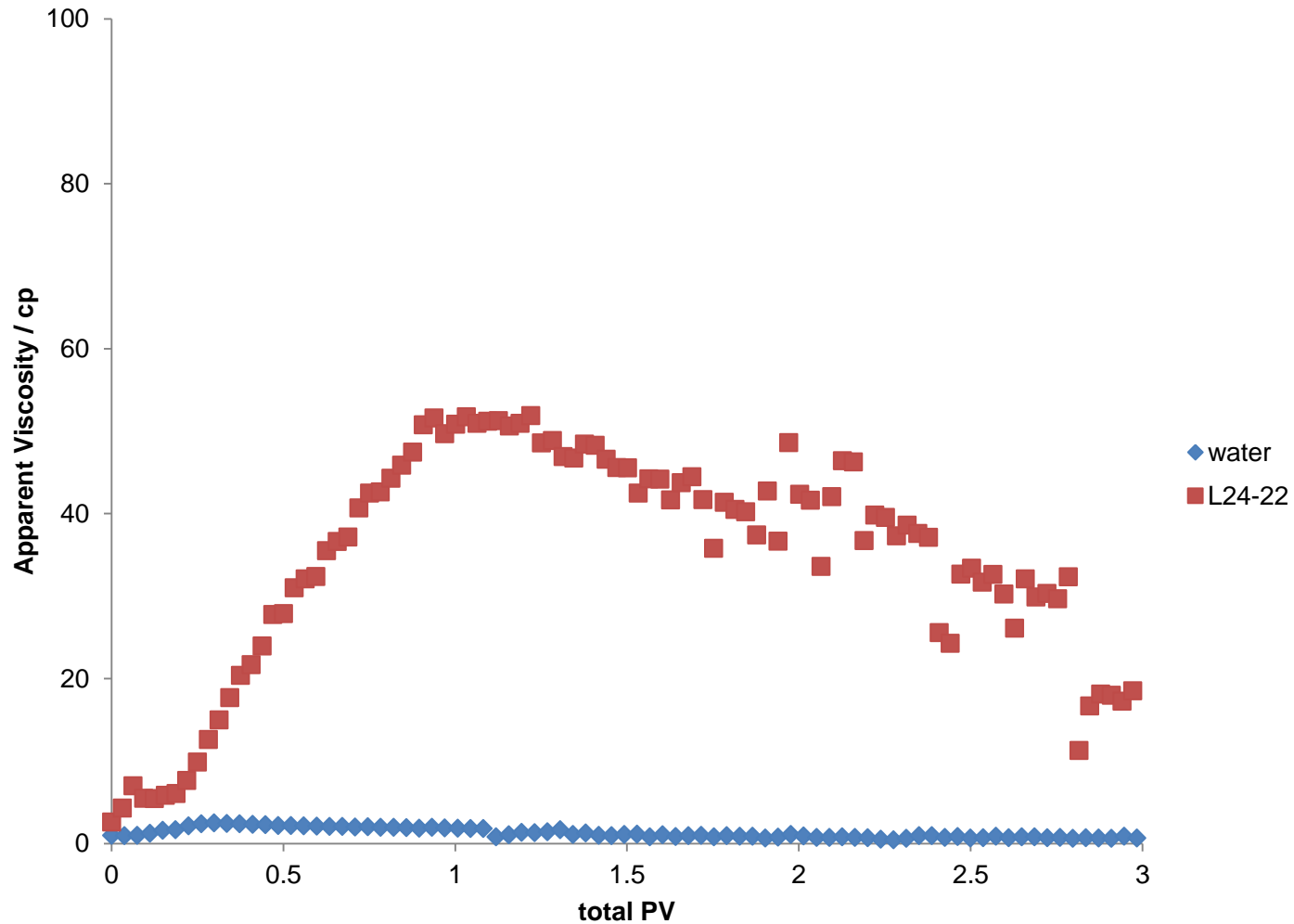
Apparent Viscosity of CO₂ Co-Injected with Water or Non-Ionic Surfactant Solution in Dolomite Core



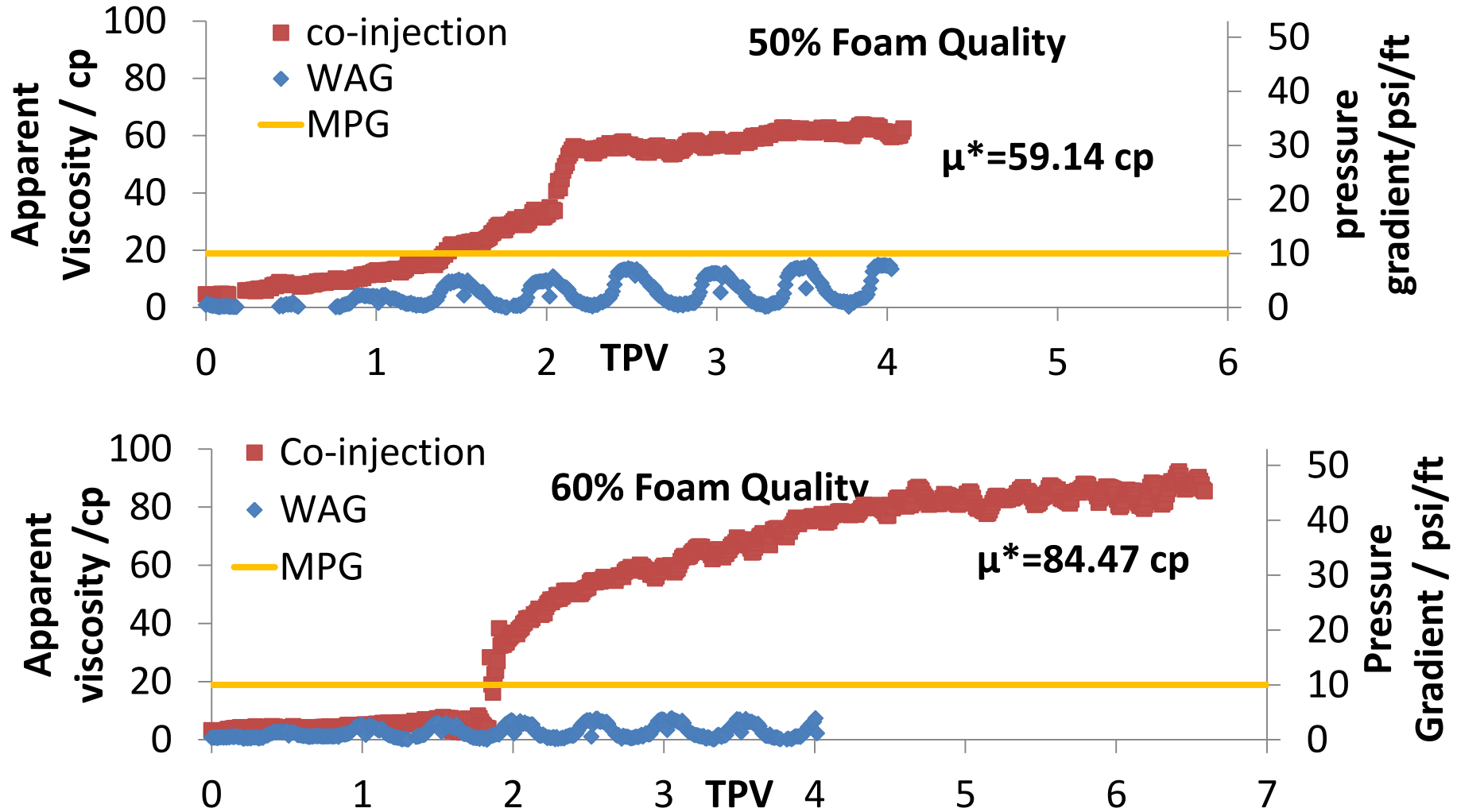
Apparent Viscosity of WAG or SAG



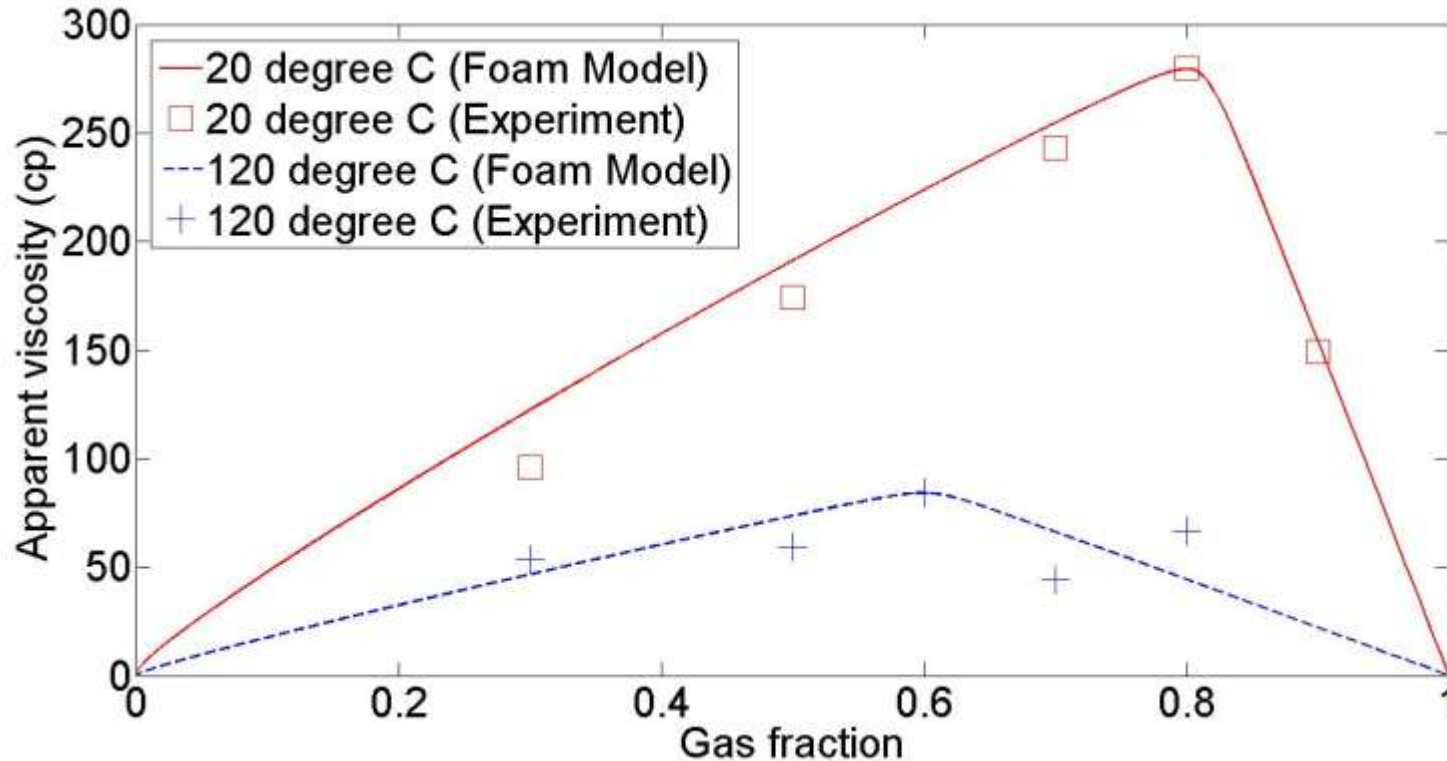
Apparent Viscosity of CO₂ Displacing Water or Non-Ionic Surfactant Solution in Dolomite Core



C12/Brine and CO₂ Foam at 120 °C and 3400 psi



Comparison of viscosity at 20 and 120 °C for C12/Brine and CO₂ Foam



$k_{rw}^0=0.5$; $k_{rg}^0=0.1768$; $S_{wc}=0.33$; $S_{gr}=0.2$; $n^w=2.8$; $n^g=1.1$; (Bennion, 2008)

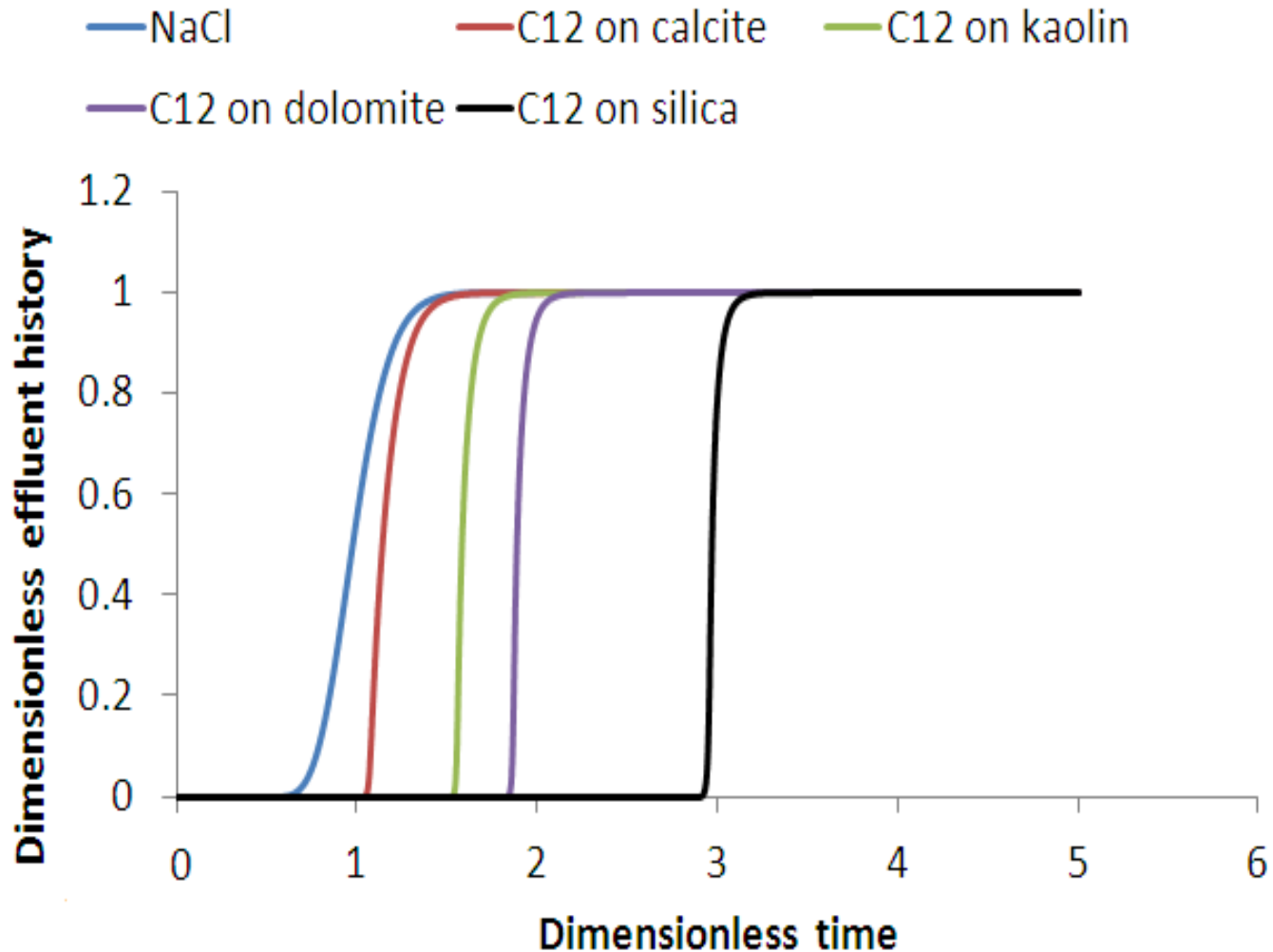
$\mu_w=0.2381$ cp; $\mu_g=3.935 \times 10^{-2}$ cp;

$epdry=10000$

Transport of surfactant is limited by adsorption on rock

- **Adsorption of anionic surfactant is generally lower on sandstone compared to carbonate**
- **Adsorption on nonionic surfactant is generally lower on carbonate compared to sandstone**
- **Nonionic surfactant has cloud-point limitation at higher temperature**
- **Cationic or nonionic/cationic surfactant has potential for carbonate formation at high temperature**

Surfactant Adsorption is Dependent on Surfactant/Mineral Interaction



Conclusions

- **CO₂ EOR can be improved by foam mobility control to improve reservoir sweep**
- **Foam improves sweep in layered systems**
- **Surfactant must be matched with reservoir parameters to limit surfactant adsorption**